R07

Code No: 07A3EC18

Set No. 2

II B.Tech I Semester Examinations, May/June 2012 PHYSICAL METALLURGY Metallurgy And Material Technology

Metallurgy And Material Technology
Time: 3 hours

Max Marks: 80

Answer any FIVE Questions All Questions carry equal marks

- 1. (a) The ionic radius of an anion is 2.11A⁰. Find the radius of the smallest cation that can have stable eight fold co-ordination with the above anions.
 - (b) Calculate the atomic packing factor of BCC structure.

[8+8]

- 2. Describe the following steps with respect to electron microscope:
 - (a) Source of illumination
 - (b) Condenser lens
 - (c) Objective lens
 - (d) Projector lens.

 $[4 \times 4 = 16]$

- 3. (a) Describe the mechanism of dendritic growth in a crystal.
 - (b) With a neat sketch explain the cooling curve for a binary solid solution alloy. [8+8]
- 4. Write short notes on the following:
 - (a) Intermediate alloy phases
 - (b) Size factor compounds

[8+8]

- 5. Discuss the different methods for obtaining a spheroidized cementite structure.[16]
- 6. Draw and explain the TTT curve of hypo eutectoid steels. Label all the phases, lines and areas and also discuss significance of each line. [16]
- 7. Write a short note on the following:
 - (a) Solid solution
 - (b) Solvus line.
 - (c) Recalescence
 - (d) Invariant point.

 $[4\times4]$

8. Discuss and draw the cooling and heating curves of pure Iron.

[16]

|R07|

Set No. 4

II B.Tech I Semester Examinations, May/June 2012 PHYSICAL METALLURGY

Metallurgy And Material Technology

Time: 3 hours Max Marks: 80

> Answer any FIVE Questions All Questions carry equal marks

> > ****

1. Write short notes on the following:

Code No: 07A3EC18

- (a) Intermediate alloy phases
- (b) Size factor compounds

[8+8]

- 2. Draw and explain the TTT curve of hypo eutectoid steels. Label all the phases, lines and areas and also discuss significance of each line. [16]
- 3. Describe the following steps with respect to electron microscope:
 - (a) Source of illumination
 - (b) Condenser lens
 - (c) Objective lens
 - (d) Projector lens.

 $[4 \times 4 = 16]$

- 4. (a) The ionic radius of an anion is 2.11A⁰. Find the radius of the smallest cation that can have stable eight fold co-ordination with the above anions.
 - (b) Calculate the atomic packing factor of BCC structure. [8+8]
- 5. Discuss the different methods for obtaining a spheroidized cementite structure. [16]
- 6. (a) Describe the mechanism of dendritic growth in a crystal.
 - (b) With a neat sketch explain the cooling curve for a binary solid solution alloy. [8+8]
- 7. Write a short note on the following:
 - (a) Solid solution
 - (b) Solvus line.
 - (c) Recalescence
 - $[4\times4]$ (d) Invariant point.
- 8. Discuss and draw the cooling and heating curves of pure Iron. [16]

R07

Code No: 07A3EC18

Set No. 1

II B.Tech I Semester Examinations, May/June 2012 PHYSICAL METALLURGY Metallurgy And Material Technology

Time: 3 hours Max Marks: 80

Answer any FIVE Questions All Questions carry equal marks

- 1. Draw and explain the TTT curve of hypo eutectoid steels. Label all the phases, lines and areas and also discuss significance of each line. [16]
- 2. Discuss the different methods for obtaining a spheroidized cementite structure. [16]
- 3. (a) The ionic radius of an anion is 2.11A⁰. Find the radius of the smallest cation that can have stable eight fold co-ordination with the above anions.
 - (b) Calculate the atomic packing factor of BCC structure.
- 4. Write short notes on the following:
 - (a) Intermediate alloy phases
 - (b) Size factor compounds

[8+8]

[8+8]

Describe the following steps with respect to electron microscope:

- 5. (a) Source of illumination
 - (b) Condenser lens
 - (c) Objective lens
 - (d) Projector lens.

 $[4 \times 4 = 16]$

- 6. (a) Describe the mechanism of dendritic growth in a crystal.
 - (b) With a neat sketch explain the cooling curve for a binary solid solution alloy.

[8+8]

- 7. Write a short note on the following:
 - (a) Solid solution
 - (b) Solvus line.
 - (c) Recalescence
 - (d) Invariant point. $[4\times4]$
- 8. Discuss and draw the cooling and heating curves of pure Iron. [16]

|R07|

Code No: 07A3EC18

Set No. 3

II B.Tech I Semester Examinations, May/June 2012 PHYSICAL METALLURGY Metallurgy And Material Technology

Metallurgy And Material Technology
Time: 3 hours

Max Marks: 80

Answer any FIVE Questions All Questions carry equal marks

- 1. Draw and explain the TTT curve of hypo eutectoid steels. Label all the phases, lines and areas and also discuss significance of each line. [16]
- 2. Write a short note on the following:
 - (a) Solid solution
 - (b) Solvus line.
 - (c) Recalescence
 - (d) Invariant point. $[4\times4]$
- 3. Write short notes on the following:
 - (a) Intermediate alloy phases
 - (b) Size factor compounds

[8+8]

Describe the following steps with respect to electron microscope:

- 4. (a) Source of illumination
 - (b) Condenser lens
 - (c) Objective lens
 - (d) Projector lens.

 $[4 \times 4 = 16]$

- 5. (a) Describe the mechanism of dendritic growth in a crystal.
 - (b) With a neat sketch explain the cooling curve for a binary solid solution alloy. [8+8]
- 6. Discuss and draw the cooling and heating curves of pure Iron. [16]
- 7. (a) The ionic radius of an anion is 2.11A⁰. Find the radius of the smallest cation that can have stable eight fold co-ordination with the above anions.
 - (b) Calculate the atomic packing factor of BCC structure. [8+8]
- 8. Discuss the different methods for obtaining a spheroidized cementite structure. [16]